

Structure-activity Relationships of Xanthocillin Derivatives as Thrombopoietin Receptor Agonist

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Abstract Xanthocillin derivatives, which show thrombopoietin receptor agonist activity, were synthesized through our developed method. Bioassay data suggest the importance of alkene geometry, the presence of substituents at the benzene ring that support hydrophobic character, and the moderate size of the molecule. One of the two isonitrile group of the natural product appears to be dispensable.

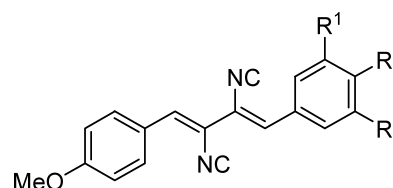
Keywords xanthocillin, structure-activity relationships, anti-viral activities, thrombopoietin receptor agonist

Thrombopoietin (TPO) is a cytokine that regulates platelet production by stimulating proliferation and differentiation of hematopoietic stem cells, megakaryocytic progenitor cells and megakaryocytes *via* activation of its receptor, c-Mpl [1]. Thrombocytopenia, or low platelet count, is one of the serious adverse events that occur as a result of intensive cancer chemotherapy or other treatments [2]. Although recombinant human (rh)TPO has been proven to be effective for such thrombocytopenia, some difficulties, such as development of neutralizing antibodies against TPO, have been reported [3]. Additionally, rhTPO and low molecular weight peptides with affinity for TPO receptor are unstable in the intestines thus limiting oral administration. Therefore, orally active, low molecular weight compounds with agonistic action on TPO receptor are in demand and several small synthetic TPO receptor

activators have been reported [4].

Xanthocillin derivatives **1**~**3** were originally isolated from cultures of soil fungus *Aspergillus* sp. in view of their anti-viral activities [5]. Sakai, R., *et al.*, however, disclosed recently that compounds **1**~**4** (Fig. 1), isolated from culture marine fungus *Basipetospora* sp., were the first natural products with TPO receptor agonist activity [6]. These compounds promoted the proliferation of a TPO-sensitive human leukemia cell line in a dose-dependent manner, and we have also reported the stereoselective synthesis of xanthocillin X dimethylether (**2**, XDE, Scheme 1) [7]. Applying our developed method, we have herein prepared several analogs to enable the study of structure-activity relationships.

Maintaining the 2,3-diisocyano-buta-1,3-diene moiety of

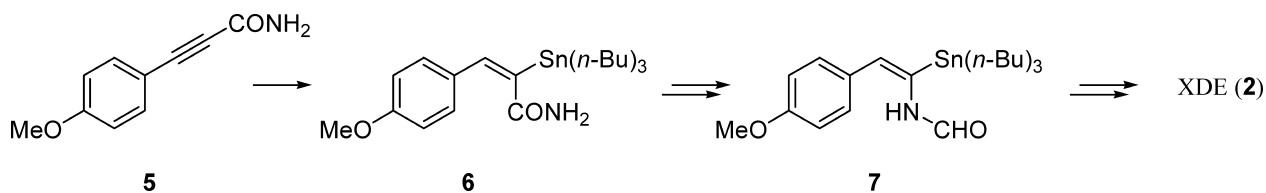


- 1:** R¹=R³=H, R²=OH
2: R¹=R³=H, R²=OMe: XDE
3: R¹=H, R²=R³=OMe
4: R¹=R²=R³=OMe

Fig. 1

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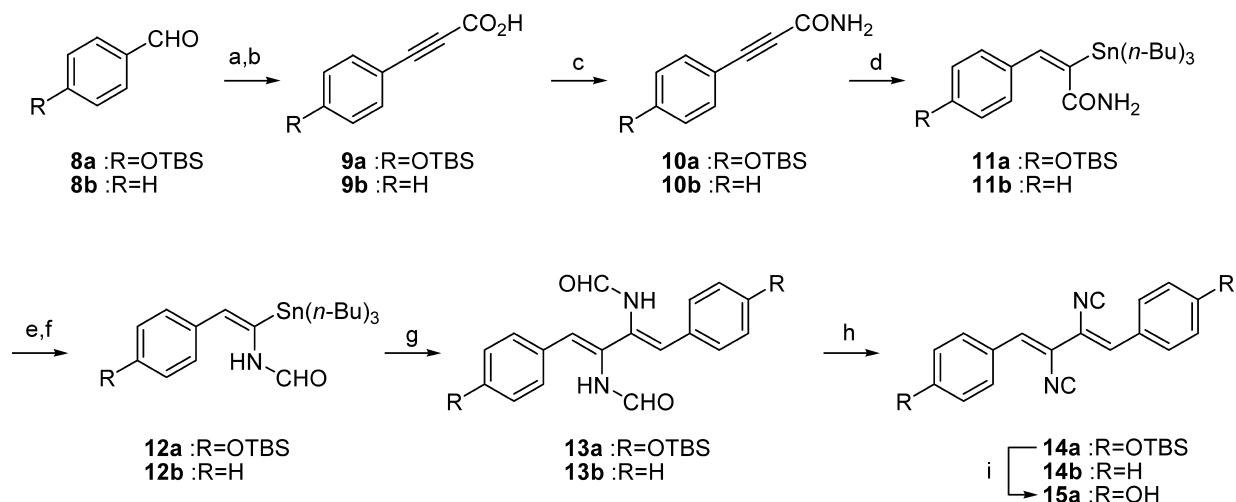
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Scheme 1.

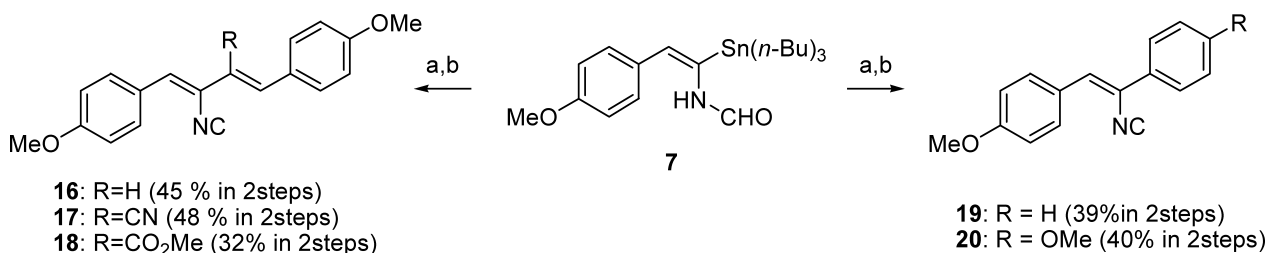
Table 1 Physico-chemical properties of bioassayed compounds

Compds.	Mp (°C) FAB-MS (<i>m/z</i>) IR (KBr) $\nu_{\text{N}=\text{C}}$ (cm^{-1})	$^1\text{H-NMR}$ (400 MHz; δ ppm) $^{13}\text{C-NMR}$ (100 MHz; δ ppm)
14b	145~147 (dec) 256 (M^+) 2114	$^1\text{H-NMR}$ (CDCl_3) 7.18 (2H, s), 7.46 (6H, m), 7.82 (4H, d, $J=8.4$ Hz). $^{13}\text{C-NMR}$ (CDCl_3) 118.1, 128.9, 129.0, 130.0, 130.5, 131.9, 174.0.
15a	135 (dec) 289 ($\text{M}+\text{H}^+$) 2134	$^1\text{H-NMR}$ (Acetone- d_6) 6.94 (4H, d, $J=8.4$ Hz), 7.04 (2H, s), 7.78 (4H, d, $J=8.4$ Hz). $^{13}\text{C-NMR}$ (Acetone- d_6) 116.9, 117.2, 125.2, 128.7, 133.2, 160.8, 175.3.
16	134 (dec) 291 (M^+) 2015	$^1\text{H-NMR}$ (CDCl_3) 3.84 (3H, s), 3.86 (3H, s), 6.43 (1H, s), 6.58 (1H, d, $J=15.2$ Hz), 6.90 (2H, d, $J=8.8$ Hz), 6.91 (1H, d, $J=15.2$ Hz), 6.95 (2H, d, $J=8.8$ Hz), 7.42 (2H, d, $J=8.8$ Hz), 7.73 (2H, d, $J=8.8$ Hz). $^{13}\text{C-NMR}$ (CDCl_3) 55.3, 55.4, 114.3, 114.3, 121.3, 128.2, 128.3, 128.4, 128.5, 128.7, 130.2, 131.0, 159.9, 159.9, 160.3.
17	170~172 (dec) 316 (M^+) 2115	$^1\text{H-NMR}$ (CDCl_3) 3.87 (3H, s), 3.88 (3H, s), 6.98 (2H, d, $J=8.8$ Hz), 6.99 (2H, d, $J=8.8$ Hz), 7.04 (1H, s), 7.48 (1H, s), 7.79 (2H, d, $J=8.8$ Hz), 7.88 (2H, d, $J=8.8$ Hz). $^{13}\text{C-NMR}$ (CDCl_3) 55.4, 55.5, 103.5, 114.6, 116.1, 117.7, 125.0, 125.3, 129.2, 131.7, 131.8, 161.1, 162.0, 173.2.
18	88~89 349 (M^+) 2111	$^1\text{H-NMR}$ (CDCl_3) 3.84 (6H, s), 3.87 (3H, s), 6.46 (1H, s), 6.90 (2H, d, $J=8.8$ Hz), 6.95 (2H, d, $J=8.8$ Hz), 7.08 (1H, s), 7.30 (2H, d, $J=8.8$ Hz), 7.72 (2H, d, $J=8.8$ Hz). $^{13}\text{C-NMR}$ (CDCl_3) 53.1, 55.5, 55.6, 114.4, 114.5, 125.8, 127.1, 127.1, 128.1, 130.3, 131.4, 131.7, 160.5, 160.9, 168.1, 168.6.
19	86 235 (M^+) 2109	$^1\text{H-NMR}$ (CDCl_3) 3.87 (3H, s), 6.94 (1H, s), 6.98 (2H, d, $J=8.8$ Hz), 7.39 (1H, t, $J=8.0$ Hz), 7.44 (2H, d, $J=8.0$ Hz), 7.67 (2H, d, $J=8.0$ Hz), 7.80 (2H, d, $J=8.8$ Hz). $^{13}\text{C-NMR}$ (CDCl_3) 55.4, 114.2, 121.2, 125.0, 125.9, 126.6, 128.8, 128.9, 131.2, 133.7, 160.4, 169.5.
20	91 265 (M^+) 2109	$^1\text{H-NMR}$ (CDCl_3) 3.86 (3H, s), 3.86 (3H, s), 6.82 (1H, s), 6.95 (2H, d, $J=8.8$ Hz), 6.97 (2H, d, $J=8.8$ Hz), 7.59 (2H, d, $J=8.8$ Hz), 7.76 (2H, d, $J=8.8$ Hz). $^{13}\text{C-NMR}$ (CDCl_3) 55.4, 55.4, 114.2, 114.2, 121.0, 124.8, 126.2, 126.3, 126.4, 130.9, 160.1, 160.3, 169.1.
24	62 210 (M^+) 2129	$^1\text{H-NMR}$ (CDCl_3) 3.87 (3H, s), 7.02 (2H, d, $J=8.8$ Hz), 7.33 (1H, ddd, $J=2.4, 6.8, 7.2$ Hz), 7.42 (2H, m), 7.46 (2H, d, $J=8.8$ Hz), 7.47 (1H, m). $^{13}\text{C-NMR}$ (CDCl_3) 55.3, 114.0, 124.3, 127.6, 127.8, 129.3, 130.2, 130.4, 138.5, 159.7, 166.3.
28	113 316 ($\text{M}+\text{H}^+$) 2121	$^1\text{H-NMR}$ (CDCl_3) 3.87 (3H, s), 3.88 (3H, s), 7.01 (2H, d, $J=8.8$ Hz), 7.03 (2H, d, $J=8.8$ Hz), 7.44 (1H, d, $J=8.0$ Hz), 7.49 (2H, d, $J=8.8$ Hz), 7.54 (2H, d, $J=8.8$ Hz), 7.61 (1H, dd, $J=2.0, 8.0$ Hz), 7.64 (1H, d, $J=2.0$ Hz). $^{13}\text{C-NMR}$ (CDCl_3) 55.3, 55.4, 114.0, 114.5, 125.8, 127.6, 128.0, 129.0, 130.2, 130.8, 131.1, 136.4, 137.4, 140.5, 159.7, 159.8, 166.2.



Scheme 2.

Conditions: a) PPh_3 , CBr_4 , Et_3N , CH_2Cl_2 , 0°C , 1 hour; b) $n\text{-BuLi}$ (2 eq), THF, -78°C to 0°C , 1 hour, then dry ice, -78°C to 0°C , 10 minutes, **9a**: 81%, **9b**: 58% in 2 steps respectively; c) ClCO_2Et , Et_3N , THF, 0°C , 30 minutes then aq. NH_3 , 0°C , 5 minutes, **10a**: 76%, **10b**: 82%; d) $n\text{-Bu}_3\text{SnH}$, $\text{Pd}(\text{PPh}_3)_4$, THF, 0°C , 30 minutes, **11a**: 65%, **11b**: 51%; e) $\text{Pb}(\text{OAc})_4$, THF, rt, 20 minutes; f) LiEt_3BH , THF, -78°C , 30 minutes, **12a**: 86%, **12b**: 83% in 2 steps respectively; g) CuCl , O_2 , THF, 0°C , 4 hours, **13a**: 50%, **13b**: 37%; h) triphosgene, Et_3N , CH_2Cl_2 , 0°C , 1 hour, **14a**: 72%, **14b**: 50%; i) TBAF, AcOH, THF, 0°C to rt, 3 hours, **15a**: 85%.



Scheme 3.

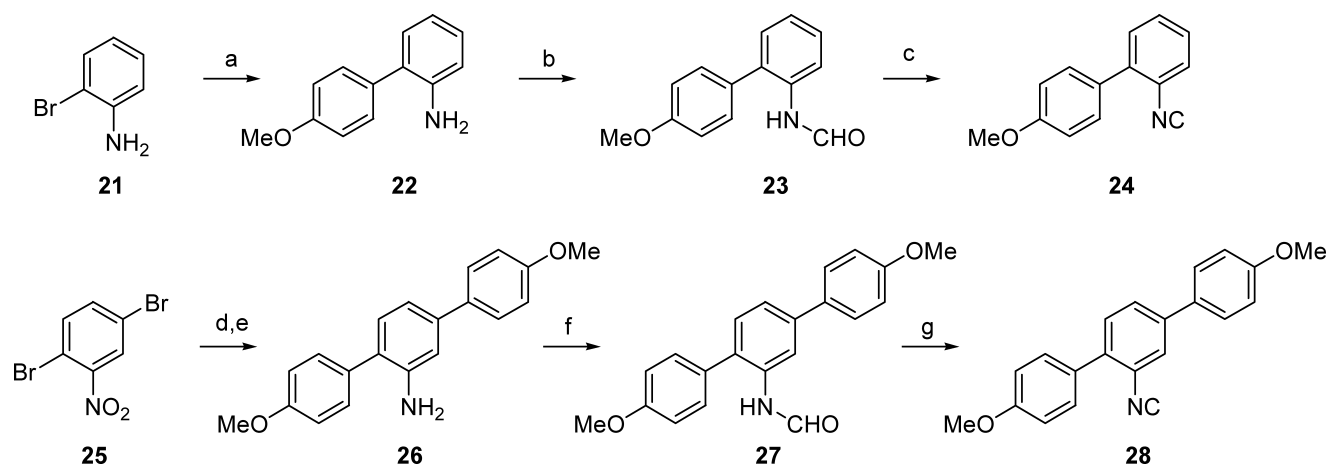
Conditions: a) ArX , $\text{PdCl}_2(\text{PPh}_3)_2$, CuI , DMF; b) triphosgene, Et_3N , CH_2Cl_2 .

the bioactive compounds, we first modified the substitution on the benzene rings. Xanthocillin X (**15a**), isolated from *Penicillium* sp. as an antibiotic in 1950 [8], and a deoxy analog **14b** were synthesized from protected 4-hydroxybenzaldehyde and benzaldehyde respectively as previously reported (Scheme 2) [7]. The benzaldehydes **8a-b** were converted to propiolic acids **9a-b** through the dibromoolefins. The mixed anhydrides of **9a-b** were treated with ammonia to give amides **10a-b**, which afforded the (*E*)-vinylstannanes **11a-b**. Successive oxidative rearrangement with $\text{Pb}(\text{OAc})_4$ proceeded to the isocyanates which were reduced with LiEt_3BH to the *N*-formyl-enamides **12a-b**. They were converted to the (*Z,Z*)-dienes **13a-b** by homocoupling with CuCl under oxygen atmosphere, which were dehydrated with triphosgene to the (*Z,Z*)-diisocyanide **14a-b**. Finally, removal of TBS groups with TBAF and acetic acid converted **14a** to xanthocillin X

(**15a**).

Mono-isocyanide compounds **16**–**20** were obtained through Stille coupling of the key intermediate **7** [7] and various aryl halides, followed by dehydration (Scheme 3). The biphenyl compound **24** and the terphenyl compound **28** were prepared *via* the substituted *N*-formanilide **23** and **27**, which were obtained through Suzuki-Miyaura coupling with *p*-methoxyboronic acid (Scheme 4).

The activities of compounds **14b**–**20**, **24**, **28** and **29** in cell proliferation of UT-7/EPO-mpl were evaluated (Fig. 2~4) [9]. The synthetic XDE (**2**) displayed the same potency as the natural **2**, although the (*E,E*)-isomer **29**, stereoselectively synthesized as previously reported [7], was inactive (Fig. 2). Unexpectedly, xanthocillin X (**15a**, with OH groups) had no agonistic potency, while xanthocillin X monomethylether (**1**) had been reported to be an active compound as well as XDE (**2**) [6]. The deoxy



Scheme 4.

Conditions: a) *p*-Methoxyphenylboronic acid, Pd(PPh₃)₄, Na₂CO₃, DMF-H₂O, 60°C, 17 hours, quant. b) acetic formic anhydride, CH₂Cl₂, rt, 1 hour, 68%; c) triphosgene, Et₃N, CH₂Cl₂, 0°C, 30 minutes, 99%. d) *p*-Methoxyphenylboronic acid, Pd(PPh₃)₄, TBAC, KOH, THF, 60°C, 3 hours, 77%; e) NaBH₄, NiCl₂·6H₂O, MeOH, 0°C to rt, 1 hour, 56%; f) acetic formic anhydride, CH₂Cl₂, rt, 30 minutes, 69%; g) Burgess reagent, THF, rt, 30 minutes, 93%.

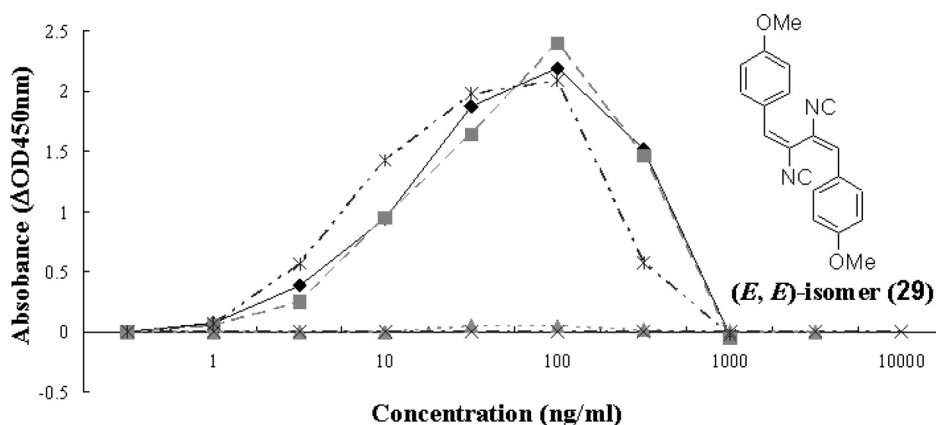


Fig. 2 UT-7/EPO-mpl cell proliferation.

◆, XDE (natural); ■, XDE (synthetic); ▲, (*E,E*)-isomer (**29**); ×, Xanthocillin X (**15a**); ✱, **14b**.

analog **14b** was as potency as XDE (**2**). These results suggested that the configuration of (*Z,Z*)-diene is essential for activity and that hydrophilic substituents on the benzene rings augment the biological activity.

Removing one of isonitrile groups did not destroy activity but reduce efficacy (compound **16**, Fig. 2). Replacement of an isonitrile with a nitrile group maintained efficacy (compound **17**), but the methylester **18** was inactive. These results indicated that two isonitrile groups are not essential; a bulky substituent, even a methylester group, however, has a deleterious effect. 1,2-Diphenylethene derivatives **19** and **20** were active, but the efficacies were lower in comparison with **2** (Fig. 3). In

contrast to the 1,4-diene derivatives, both compounds were light-stable. The biphenyl derivative **24** was inactive, while the terphenyl compound **28**, which is also light-stable, exhibited efficacy comparable to **2**. Considering the length of the molecules, **24** is too compact to be active, and moderate size is essential for activity.

In conclusion, these studies demonstrated that 1) the olefin geometry affects the activity, 2) the activity benefits by hydrophobic substituents on benzene rings, 3) two isonitrile groups are not needed to sustain biological activity, and 4) the moderate size of molecule seems to be important. We found the deoxy analog **14b** to show equivalent activity, and stable compounds, **20** and **28**, to

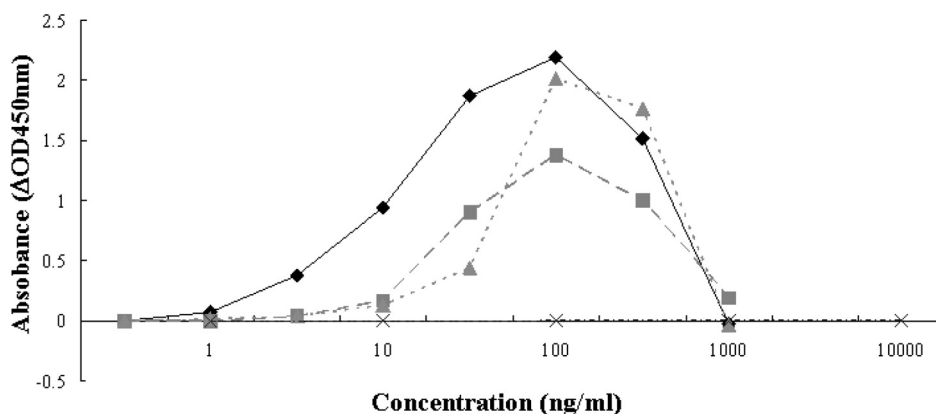


Fig. 3 UT-7/EPO-mpl cell proliferation
◆, XDE (natural); ■, 16; ▲, 17; ×, 18.

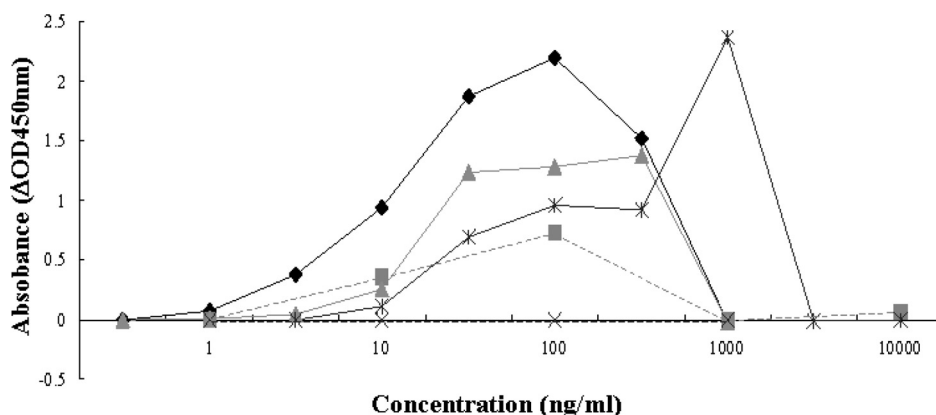


Fig. 4 UT-7/EPO-mpl cell proliferation
◆, XDE (natural); ■, 19; ▲, 20; ×, 24; ⋈, 28.

maintain the activity. Regardless of the variety of substituents, all active compounds showed inhibition of cell proliferation at high concentration. This effect may be due to the inherent cytotoxicity of the vinyl isonitrile group which unfortunately appears to be indispensable for the activity.

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 9. UT-7/EPO-mpl (kindly donated by Dr Norio Komatsu, University of Yamanashi) was prepared by introducing a vector that induces expression of human TPO receptor gene (c-mpl) into human leukemia cell line UT-7/EPO [10]. Proliferation of UT-7/EPO-mpl is stimulated by TPO, while its parental cell line UT7/EPO exhibits no response to TPO. The activities of the compounds on the cell proliferation of UT-7/EPO-mpl cell were evaluated according to the reported procedure [5]. Briefly, the cells were subcultured in Iscove's modified Dulbecco's medium (IMDM; Gibco) containing 10% fetal bovine serum (FBS; BioWest) using a CO₂ incubator (5% CO₂, 37°C). The subcultured cells were washed twice with phosphate buffered saline (PBS) and suspended in IMDM containing 10% FBS at a cell density of 6×10⁴ cells/ml. The cell suspension was transferred to a 96-well tissue culture plate in 100-μl aliquots. Then either TPO (PeproTech EC) or the compound dissolved in DMSO was diluted 83-fold with IMDM containing 10% FBS and added to the aforementioned cell suspension in 20-μl aliquots. The suspension was incubated in a CO₂ incubator (5% CO₂, 37°C) for 4 days. Cell proliferation was assayed using WST-8 reagent (Kishida Chemical) according to instructions by the manufacturer. A 10-μl aliquot of 5 mM WST-8 reagent was added to each well of the tissue culture plate, and the plate was incubated at 37°C for 4 hours. The formazan pigment generated was detected by measuring the absorbance at 450 nm with a 96-well microplate reader (Nihon Molecular Devices, Spectramax 190)
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